

# Global Biogeochemical Cycles®

## RESEARCH ARTICLE

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### Key Points:

- Compared to the dry season, dam operation increased the riverine dissolved inorganic carbon (DIC), and lake recharge decreased the DIC in the wet season
- Water retention by the reservoir increased the carbonate dissolution proportion of riverine DIC, but lakes flow mixed into the river decreased it
- Sediment mineralization increased annual  $p\text{CO}_2$  and DIC concentrations in the Three Gorges Reservoir

### Supporting Information:

Supporting Information may be found in the online version of this article.

### Correspondence to:

C. Huang,  
[huangchangchun\\_aaa@163.com](mailto:huangchangchun_aaa@163.com)

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### Author Contributions:

**Writing – original draft:** Chu Zhao  
**Writing – review & editing:** Chu Zhao

## Changes in Dissolved Inorganic Carbon Across Yangtze River Regulated by Dam and River-Lake Exchange

Chu Zhao<sup>1</sup>, Chuan Wang<sup>1</sup>, Jianhong Li<sup>1,2</sup>, Lize Meng<sup>1</sup>, Jingya Xue<sup>1</sup>, Yang Gao<sup>3</sup> , Tao Huang<sup>1</sup>, Yixin Bai<sup>1</sup>, Shuaidong Li<sup>1</sup>, Hao Yang<sup>1</sup>, Kun Shi<sup>4</sup> , Yuanhui Xu<sup>1</sup>, and Changchun Huang<sup>1</sup> 

<sup>1</sup>School of Geography, Nanjing Normal University, Nanjing, PR China, <sup>2</sup>Key Laboratory of Karst Dynamics, MNR & Guangxi, Institute of Karst Geology, Chinese Academy of Geological Sciences, Guilin, PR China, <sup>3</sup>Key Laboratory of Ecosystem Network Observation and Modeling, Institute of Geographic Sciences and Natural Resources Research, CAS, Beijing, PR China, <sup>4</sup>Taihu Laboratory for Lake Ecosystem Research, State Key Laboratory of Lake Science and Environment, Nanjing Institute of Geography and Limnology, Chinese Academy of Sciences, Nanjing, PR China

**Abstract** The boom in dam construction and continuous river-lake exchange has had a profound impact on the transmission and transformation of riverine dissolved inorganic carbon (DIC). An in-depth understanding of the change mechanisms of DIC concentrations and sources driven by dam operation and lake recharge is crucial for regulating greenhouse gas emissions and evaluating the impact of DIC on the global carbon cycle. This study investigated dam- and lakes-driven DIC via the concentration and  $\delta^{13}\text{C}$  of DIC, combined with anions, cations,  $\delta\text{D}$  and  $\delta^{18}\text{O}$  in the main stream of the Yangtze River. DIC showed a decreasing trend from upper reach ( $2,262.31 \pm 113.69 \mu\text{mol kg}^{-1}$ ) to lower reach ( $1,771.61 \pm 89.36 \mu\text{mol kg}^{-1}$ ). Carbonate dissolution proportion (from 36.45% to 28.44%) and atmospheric  $\text{CO}_2$  proportion (from 37.51% to 22.94%) of DIC decreased from upper reach to lower reach, whereas soil  $\text{CO}_2$  proportion of DIC (from 26.01% to 48.63%) increased. The control of dam operation on DIC biogeochemical process was revealed from different time scales. From the perspective of short-term seasonal changes (from 2020 to 2021), the mineralization of organic matter in the dry season strengthened  $\text{CO}_2$  degassing and calcite precipitation, reducing the DIC and increasing the proportion of soil  $\text{CO}_2$ . Meanwhile, longer periods of runoff retention provided sufficient time for water–rock reactions in the wet season and increased the DIC and carbonate dissolution source in the reservoir area. On a long-term scale (from 2009 to 2021), a decrease in pH driven by sediment mineralization contributed to an annual increase in DIC in the reservoir. The flow of lakes mixed into the mainstream was revealed by the enrichment of  $\delta^{18}\text{O}$ , and river-communicating recharge decreased the DIC and carbonate dissolution source. We show that dam operation and lake inflow change DIC concentrations and sources and therefore need to be considered in the transmission and transformation processes of DIC in the river-ocean continuum.

## 1. Introduction

Rivers are vinculum of material connections between the atmosphere, soil, and biosphere, which contribute to the export of terrestrial carbon into estuaries and oceans and play a significant role in the global carbon cycle (Aufdenkampe et al., 2011; Catalán et al., 2016; Mendonça et al., 2017). Dissolved inorganic carbon (DIC) is an integral part of the carbon pool and is affected by the processes of photosynthesis, organic matter decomposition, calcite precipitation and dissolution that occur in the river system. Thus, mass transport and transformation of DIC have profound impacts on the global carbon cycle (Gao et al., 2022; Isaji et al., 2017; Pu et al., 2017). The process of  $\text{CO}_2$  evasion from rivers is an important transformation of riverine DIC (Akhand et al., 2021; Xu et al., 2019) because most rivers are supersaturated with  $\text{CO}_2$  (Raymond et al., 2013). Previous studies have proposed that the export of carbon from terrestrial ecosystems to inland water reached 1.9 Pg carbon per year ( $\text{PgC yr}^{-1}$ ), of which 0.43 PgC was transported from rivers to the ocean in the form of DIC, and 0.8 PgC was transformed to  $\text{CO}_2$  then emitted from rivers to the atmosphere (Cole et al., 2007; Probst et al., 1994; van Hoek et al., 2021). During the past decade, export of carbon from terrestrial ecosystems to inland water increased to 5.1  $\text{PgC yr}^{-1}$ . Anthropogenic activities and organic matter mineralization increase DIC by varying degrees in many large rivers, such as Yangtze River, Ob River, and Mississippi River, in which the DIC concentrations have increased by 28.82%, 51.43%, and 61.23%, respectively (Cai et al., 2015; Drake et al., 2018; Gordeev et al., 1996; Hearth et al., 2022; Pipko et al., 2019; Voss et al., 2017; Zhang et al., 2014). Import of terrestrial carbon into rivers increased soil  $\text{CO}_2$  proportion of DIC, thus  $\text{CO}_2$  in most rivers becomes more supersaturated and contributes to global riverine  $\text{CO}_2$  emissions, which have increased to 3.28  $\text{PgC yr}^{-1}$  (Marx et al., 2017; Wen et al., 2021). Therefore, source

proportion and transformation processes of DIC have received increasing attention in geochemistry and with respect to the global carbon cycle (Campeau & Del Giorgio, 2014; Raymond & Hamilton, 2018).

Concerns regarding riverine DIC are primarily due to the substantial changes in its concentration caused by comprehensive natural and anthropogenic factors (Lu et al., 2022; Papadimitriou et al., 2005; Zhang et al., 2020). In natural water,  $\text{HCO}_3^-$  predominates DIC, which is partially regulated by the availability of weatherable minerals and the supply of weathering agents; thus, exogenous acids from agricultural and industrial production increase riverine DIC. Land leveling imports terrestrial materials into the river and promotes the contribution of organic matter mineralization to the increase in DIC. Dams intercept runoff and silt and regulate the riverine DIC's transport and transformation processes. The variation in riverine recharge will lead to an enrichment or dilution effect on DIC and may further affect the source proportion of DIC (Gao et al., 2020; Liu et al., 2016; Raymond & Hamilton, 2018; Song et al., 2020). Among these influential factors, the effect of damming and lake recharge on DIC has aroused widespread concern (Li, Pu, & Zhang, 2022; Wang et al., 2020). As over 16 million dams have been operating globally and many new dams will be built, biogeochemical processes of riverine DIC have significantly changed (Maavara et al., 2017). Heterotrophic microorganisms decompose terrestrial matter intercepted by dams to increase DIC and  $\text{CO}_2$  proportion.  $\text{CO}_2$  supersaturation results in outgassing and calcite precipitation to remove  $\text{CO}_2$ , then calcite precipitates and forms the inorganic carbon sink of the sediment (Gao et al., 2021; Zeng et al., 2019; Zhong et al., 2021a). Dams prolong the runoff retention period, reduce flow, strengthen water-rock interactions, and contribute to an increase in DIC concentration and carbonate dissolution. On the other hand, the flow of river-communicating lakes mixed into mainstream may directly change the DIC concentration and its source proportion because of different ion concentrations and rock weathering between lakes and rivers (Guo et al., 2023; Zhang et al., 2014). However, some knowledge gaps still exist regarding DIC transmission and conversion processes affected by damming and lake recharge. The seasonal variation in DIC production and consumption in reservoirs and the feedback mechanism of the response of the carbonate balance system to the continuous saturation of  $\text{CO}_2$  in water remain unclear against the background of continuous exogenous carbon input. In monsoon regions, seasonal differences in the influence of lake recharge on riverine DIC require further evaluation. Carbon, deuterium, and oxygen isotopes will undergo fractionation due to differences in water retention time and supply sources; thus, isotope tracer technology can be used to investigate the effective mechanism of dam operation and lake recharge on DIC sources (Leybourne et al., 2006; Reyes-Macaya et al., 2022).

Riverine DIC originate from potential sources such as atmospheric  $\text{CO}_2$ , carbonate dissolution, and soil  $\text{CO}_2$  (Chen et al., 2021; Song et al., 2020). The weathering of carbonate and silicate rocks consumes atmospheric  $\text{CO}_2$  to produce DIC (Abongwa & Atekwana, 2015; Chetelat et al., 2008). Half of the carbon produced by carbonate weathering comes from atmospheric  $\text{CO}_2$ , and the other half comes from self-dissolution (Gao et al., 2020; Herath et al., 2020). Biogenic DIC originates from soil  $\text{CO}_2$  produced by organic mineralization and  $\text{CO}_2$  (aqueous) formed by soil  $\text{CO}_2$  dissolved in a solution (Cai et al., 2015; Telmer & Veizer, 1999). Due to the differences in activities between light ( $^{12}\text{C}$ ) and heavy carbon isotopes ( $^{13}\text{C}$ ), carbon isotopes of DIC ( $\delta^{13}\text{C}_{\text{DIC}}$ ) undergo a series of physical, chemical, and biological reaction processes that result in different degrees of fractionation. For example, soil organic matter mineralization causes  $\delta^{13}\text{C}_{\text{DIC}}$  to be influenced by the carbon isotope characteristics of terrestrial plants, thus causing  $\delta^{13}\text{C}_{\text{DIC}}$  depletion, while atmospheric  $\text{CO}_2$  weathering and carbonate dissolution cause  $\delta^{13}\text{C}_{\text{DIC}}$  to become relatively enriched (Cotovicz et al., 2019; Samanta et al., 2015). Therefore, the proportion of DIC sources resulting from different reactions can be accurately reflected by variations in the  $\delta^{13}\text{C}_{\text{DIC}}$  values. DIC formed by carbonate dissolution typically has a  $\delta^{13}\text{C}_{\text{DIC}}$  value of  $-4.2\text{‰}$  to  $-3.6\text{‰}$  and an average of  $-3.9\text{‰}$  (Ishikawa et al., 2015). DIC originating from rock weathering consumes atmospheric  $\text{CO}_2$  with a  $\delta^{13}\text{C}_{\text{DIC}}$  of approximately  $-2.5\text{‰}$  (range  $-4.5\text{‰}$  to  $-0.5\text{‰}$ ) (Ishikawa et al., 2015; Shan et al., 2021). Soil  $\text{CO}_2$  is formed by organic matter decomposition, and its  $\delta^{13}\text{C}_{\text{DIC}}$  value (from  $-20.5\text{‰}$  to  $-18.5\text{‰}$ , average  $-19.5\text{‰}$ ) is affected by the overlying vegetation type (Telmer & Veizer, 1999). The  $\delta^{13}\text{C}_{\text{DIC}}$  value of soil  $\text{CO}_2$  dissolved in soil solution varies from  $-23\text{‰}$  to  $-13\text{‰}$  with an average of  $-17\text{‰}$  (Amiotte-Suchet et al., 1999; Telmer & Veizer, 1999). As a result, additional isotope analysis of DIC may improve our understanding of the seasonal and spatial changes in DIC and the associated control mechanisms.

In the current study, we focused on the main stream of the Yangtze River to elucidate the effects of lake recharge and operation of the Three Gorges Dam (TGD) on DIC concentration and source. We adopted methods of DIC dynamic analysis, isotope tracer technology, and redundancy analysis to: (a) clarify the spatiotemporal distribution characteristics of DIC and sources in the main stream of the Yangtze River; (b) assess synchronous changes

in DIC concentration and sources disturbed by reservoir effect and river-communicating lakes recharge; and (c) reveal the environmental factors that affect the concentration and source of DIC.

## 2. Material and Methods

### 2.1. Study Area and Sampling

The Yangtze River originates from the Qinghai-Tibet Plateau and flows into the East Sea of China, with a main stream length of 6,400 km covering an area of  $1.8 \times 10^6$  km<sup>2</sup> (Wang et al., 2008). The rock formation of the Yangtze River is mainly composed of Precambrian–Quaternary alluvium, evaporites, and carbonates (Zhao et al., 2022). Carbonate rocks are widely spread over the entire basin, particularly in the upper part of the main stream (Figure S1 in Supporting Information S1). The Dongting and Poyang Lakes are two river-communicating lakes mainly composed of sedimentary and metamorphic rocks located 1,400 and 800 km from the estuary, respectively (Zhang et al., 2014). The Yangtze River Basin is subject to a subtropical monsoon climate, alternately controlled by the Siberian winter and East Asian summer monsoon. The mean annual precipitation in the watershed is ~1,142 mm. Precipitation during the wet season (May–October) accounts for 70%–90% of the total yearly rainfall (Wu et al., 2022; Zhang et al., 2014). Precipitation patterns during the two periods in this study showed significant differences; the monthly mean precipitation in December 2020 (14.1 mm) was lower than that in May 2021 (123.8 mm) (Figures S2a and S2b in Supporting Information S1). In main stream of the Yangtze River, monthly average discharge observed by hydrological stations in May ( $2.07 \times 10^{10}$  m<sup>3</sup> S<sup>-1</sup>) is significantly higher than that in December ( $4.55 \times 10^{10}$  m<sup>3</sup> S<sup>-1</sup>) (Figure S2c in Supporting Information S1). Therefore, December is considered as dry season and May is wet season in this study.

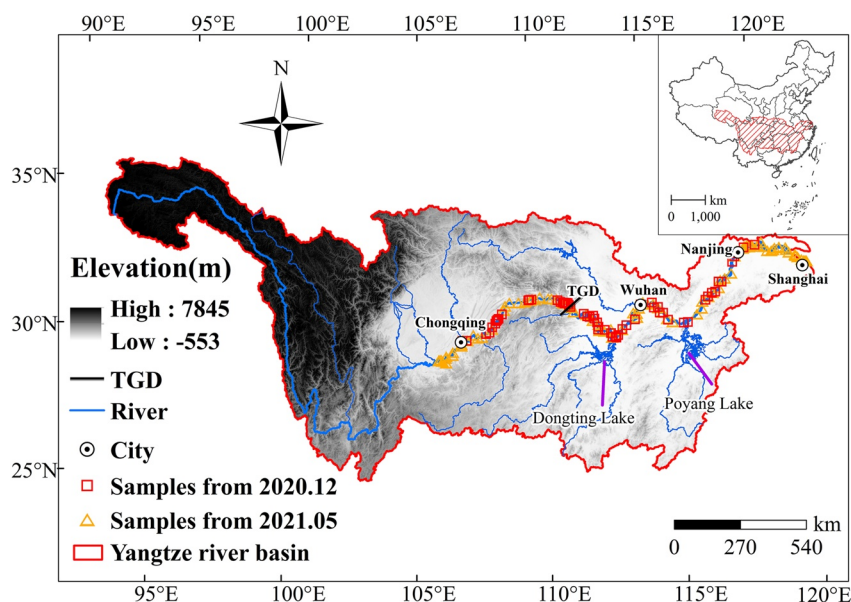
The TGD is located in the upper reaches of the Yangtze River, which extends from 1,850 to 2,300 km from the estuary of the East China Sea. The upstream of the TGR is more than 2,300 km away from the estuary, and the downstream of the TGD is 1,400–1,850 km away from the estuary. The operation state of the TGD can be divided into three stages: the construction stage (1994–2003), initial operation stage (2003–2008), and normal operation stage (after 2008) (Deng et al., 2016; Yang et al., 2014). In this process, the capacity and water regulation effect of the reservoir were enhanced, and its regulatory impact on the hydrology of the main stream was gradually highlighted. The maximum water level and storage capacity can reach 175 m and 39.3 km<sup>3</sup>, respectively (Guo et al., 2021).

During the cruise from Chongqing to Shanghai along the main stream of the Yangtze River, 164 surface water samples were collected in December 2020 (62 samples collected within 14 days) and May 2021 (102 samples collected within 28 days) using a 10 L Niskin bottle (Figure 1). Subsequently, all water samples were filtered through 0.7 μm pre-combusted (450°C for 6 hr) Whatman quartz fiber filters. The filtrate was stored into pre-cleaned high-density polyethylene (HDPE) bottles and stored in the dark at 4°C, then samples were transported to laboratory immediately.

### 2.2. Sampling and Laboratory Analyses

#### 2.2.1. Chemical Analysis of Water Samples

The total dissolved carbon (TDC), dissolved organic carbon (DOC) in water were measured using a carbon elemental analyzer (Shimadzu Corporation., Japan) (Huang et al., 2018). Total suspended particulate matter (TSM) was measured by filtration with suspended particles dried at 55°C for 8 hr (Zhao et al., 2021). The IC-free TSS and filtrate were also used to measure particulate organic carbon (POC) using a carbon elemental analyzer (Huang et al., 2022). A multi-parameter meter (YSI ProDss., USA) was used to analyze the water pH (scale is 0–14), dissolved oxygen (DO), specific conductivity (SpC), turbidity, and water temperature (Tw) for field measurements. Chlorophyll-*a* (Chl *a*) and total nitrogen (TN) levels were determined using a UV-3600 spectrophotometer (Shimadzu Corp., Japan) (Huang et al., 2017). Major anions and cations (Ca<sup>2+</sup>, Mg<sup>2+</sup>, SO<sub>4</sub><sup>2-</sup>, Na<sup>+</sup>) were measured using Dionex ICS-900 ion chromatography (Dionex, USA) based on the APHA 2012 method (Rice et al., 2012). Acid neutralizing capacity (ANC) was calculated by anions and cations. Saturation index of calcite (SIc) was calculated by the software of PHREEQC Interactive 3.6.2 (Text S1 in Supporting Information S1). δD and δ<sup>18</sup>O were measured using a mouse FAT/LEAN imaging analyzer (Picarro, USA) at the Chinese Academy of Geological Sciences (Guangxi, China) (Wu et al., 2022), and deuterium excess (d-excess) was calculated using the formula: d-excess = δD – 8δ<sup>18</sup>O (Craig, 1961).



**Figure 1.** Study area and sampling sites. The red squares and orange triangles represent water samples collected in December 2020 and May 2021, respectively. The location of TGD is shown in the figure, together with the metropolitan cities Chongqing, Wuhan, Nanjing, and Shanghai.

### 2.2.2. Measurement of DIC Concentration and Its Stable Carbon Isotope

DIC concentration were measured using TOC-L analyzer (Shimadzu Corporation., Japan) equipped with an ASI-Lsampler in total inorganic carbon (IC) mode (Gu et al., 2020; Wang et al., 2016). We pretreated water samples using glass fiber filters (Whatman quartz fiber filters, 0.7  $\mu\text{m}$ ) to avoid interference caused by PIC (particulate inorganic carbon, mean value of 22.09  $\mu\text{mol kg}^{-1}$ , Text S2 and Figure S3 in Supporting Information S1) on the DIC and carbon isotope analysis. To evaluate the influence of air-water exchange on the precision and accuracy of DIC during filtration process, we used CO<sub>2</sub>SYS software to calculate DIC through total alkalinity (TA), Tw and pH (Pierrot et al., 2006). The TA was determined by unfiltered water in the field using hydrochloric acid (HCL) titration with a Titrette Digital Titrator (Brand Trading Co., Ltd, Wertheim, Germany), HCL titrant was 0.05 mmol L<sup>-1</sup> (Zhang et al., 2019). The comparison result showed precision error between measured and calculation value was 5.73% (in December) and 4.77% (in May) (Figure S4 in Supporting Information S1), we used measured DIC by TOC-L analyzer in this study. Isotopes of DIC ( $\delta^{13}\text{C}_{\text{DIC}}$ ) were determined using Thermo MAT253 plus (Thermo Fisher Scientific, Germany) with a Gas Bench II automated device (Thermo Fisher Scientific, Germany) (Su et al., 2019). The carbon isotope ratios of DIC were expressed in delta notation ( $\delta^{13}\text{C}$ ) relative to Vienna PDB, and the standard deviations based on replicate measurements for  $\delta^{13}\text{C}$  were  $\sim 0.05\text{‰}$  (Li, Pu, & Zhang, 2022).

### 2.3. Model Application

#### 2.3.1. Bayesian Mixing Model

DIC source analysis by a Bayesian mixing model was conducted using the “MixSIAR” package in R 4.0.4 software, which is based on the Dirichlet distribution and the logical prior distribution of the Bayesian framework (Duvert et al., 2020; Evans et al., 2000). The Bayesian mixing model improves on the simpler linear hybrid model by explicitly considering uncertainty in source values, categories, and continuous covariates (more described in Text S3 of the Supporting Information S1). Previous experiments illustrated that  $\delta^{13}\text{C}_{\text{DIC}}$  does not exhibit significant changes in 10 hr (Abongwa & Atekwana, 2018). However, numerous reservoirs in the main stream of the Yangtze River contribute to the difference in the hydraulic retention time in stream segments, which leads to significant challenges and uncertainty in the calculation of DIC isotope fractionation caused by the water-atmosphere exchange in the large river. Accordingly, the DIC in the main stream of the Yangtze River was considered to be in chemical equilibrium with degassed CO<sub>2</sub>. Bayesian mixing models were not involved



in isotope fractionation caused by water-atmosphere CO<sub>2</sub> interaction (Shan et al., 2021; Song et al., 2021). Data from the δ<sup>13</sup>C<sub>DIC</sub> were categorized based on each month's upper, middle, and lower reaches to calculate the source component. The range of DIC sources was statistically estimated using end-member values from previous studies (Table S1 in Supporting Information S1). In addition, the δ<sup>18</sup>O values of the main stream before lake recharge and of the lake (Table S3 in Supporting Information S1) were viewed as two end members, based on the variation of δ<sup>18</sup>O after lake recharge into the main stream, and were to calculate the contribution of the lake water supply to the main stream of the Yangtze River.

### 2.3.2. Structural Equation Modeling

Structural equation modeling (SEM) was used to test and quantify the impact of environmental factors (temperature and pH that affect carbonate equilibrium system) and dam behavior on the DIC transformation process in the reservoir area. SEM is an advanced generalized linear regression model, which evaluates the effectiveness of covariance matrix by computing multiple relationships between variable related to DIC transformation (Amini & Alimohammadlou, 2021). The model can not only examine the direct influence among variables but also reveal the indirect influence among variables. Important parameters involved in the model are goodness of fit index (GFI) and comparative fit index (CFI), which should be more than 0.90. Root mean square error of approximation (RMSEA) should be less than 0.10 to illustrate reasonable and reliable network (Nakayama et al., 2022). In this study, GFI and CFI of the model are more than 0.95, RMSEA is less than 0.05; thus, the credibility of the model has been verified. The models were designed using the software of AMOS 25.0.

### 2.4. Data Sources Collected From Published Data

To assess long-term variation of DIC concentration affected by TGR and lakes, historical data sets including discharge, sediment loads, water quality parameters, δD, δ<sup>18</sup>O, pCO<sub>2</sub>, concentrations of riverine nutrients and elements were gathered. Discharge and sediment records were collected from previous study (Zhang et al., 2014) and the Yangtze Sediment Bulletin (<http://www.cjw.gov.cn/>), respectively. Water quality parameters δD, δ<sup>18</sup>O, pCO<sub>2</sub>, nutrients, and major ions were extracted from published sources given in Tables S3–S6 of the Supporting Information S1.

### 2.5. Processes Affecting DIC Concentrations and δ<sup>13</sup>C<sub>DIC</sub>

The variation in the DIC concentration and δ<sup>13</sup>C<sub>DIC</sub> in most reservoirs is often caused by a mixture of biogeochemical and physical processes, which include (a) CO<sub>2</sub> exchange across the air-water interface, (b) photosynthesis dominated by phytoplankton, (c) degradation of organic matter, and (d) dissolution or precipitation of calcite (Samanta et al., 2015; Su et al., 2019). To reveal the relevant processes and driving factors of DIC migration and transformation, water in the upstream of TGR was used as the reference value to evaluate DIC transformation processes in the reservoir and downstream. The changing degree of DIC and δ<sup>13</sup>C<sub>DIC</sub> was calculated by following equations (Wang et al., 2019, 2020):

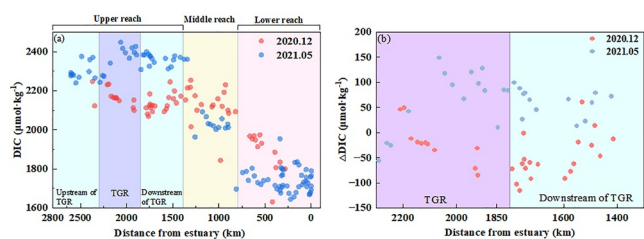
$$\Delta [\delta^{13}\text{C}_{\text{DIC}}] = \delta^{13}\text{C}_{\text{DIC}}(\text{in reservoir or downstream}) - \delta^{13}\text{C}_{\text{DIC}}(\text{upstream}) \quad (1)$$

$$\Delta [\text{DIC}] = [\text{DIC}](\text{in reservoir or downstream}) - [\text{DIC}](\text{upstream}) \quad (2)$$

## 3. Results

### 3.1. Decreasing Trend of DIC From the Upper to Lower Reaches

The concentration of DIC varied from 1,632.58 to 2,448.50 μmol kg<sup>-1</sup> with mean value of 2,049.48 ± 242.50 μmol kg<sup>-1</sup>, which displayed a gradually decreasing trend from the upper to lower reaches. DIC presented a linear relationship with the distance from the estuary (*y* defined as DIC and *x* is distance of samples from estuary, then *y* = 0.27*x* + 1,729.61, *R*<sup>2</sup> = 0.76) (Figure 2a). The seasonal difference in DIC in the main stream of the Yangtze River was significant (Table S2 in Supporting Information S1), the highest DIC was observed in December (mean value of 2,092.06 ± 126.05 μmol kg<sup>-1</sup>), followed by in May (2,023.61 ± 289.00 μmol kg<sup>-1</sup>) (Figure 2a). In addition, the DIC in May showed a higher spatial change rate (linear regression slope with samples distance



**Figure 2.** (a) Spatiotemporal variation of DIC exhibiting decrease trend from upper to lower reach of the Yangtze River. Note that the main stream was separated into upper (blue background), middle (yellow background), and lower reach (red background), the extent of TGR was highlighted by purple background; the fitting line between distance (from estuary to upper reach) and DIC during two periods was  $y = 0.27x + 1,729.61$ ,  $R^2 = 0.76$ ; the fitting line in December 2020 was  $y = 0.15x + 1,882.83$ ,  $R^2 = 0.49$  and the fitting line in May 2021 was  $y = 0.30x + 1,698.32$ ,  $R^2 = 0.83$ . (b) The difference of  $\Delta$  [DIC] between samples in TGR and downstream of TGR.

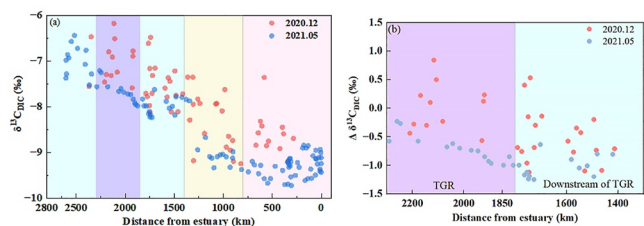
from estuary, slope = 0.30) and spatial variation (coefficient of variation, CV = 0.14) than those in December (slope = 0.15, CV = 0.06).

Cluster analysis of DIC suggested that the main stream of the Yangtze River can be separated into three reaches (upper, middle, and lower) (Figure S5 in Supporting Information S1), and the boundaries were found at 1,400 km (located near the Dongting Lake) and 800 km (located near the Poyang Lake) away from the estuary. DIC showed significant differences between each reach ( $p < 0.05$ ) (Figure 2a; Table S2 in Supporting Information S1), which in the upper reach (averaged  $2,257.02 \pm 110.39 \mu\text{mol kg}^{-1}$ ) was 8.06% and 27.70% higher than that in the middle reach and lower reach, respectively. In December, DIC decreased slightly from upper reach to middle reach, contributing to its higher value compared to the samples in May in middle reach (Figure 2a; Table S2 in Supporting Information S1). In addition, the mean value of the DIC in TGR was lower than its upstream in December, but the opposite variation of DIC was observed in May. The DIC in the TGR was higher than that in its downstream. Notably, the mean value of the DIC in TGR in May was  $201.34 \mu\text{mol kg}^{-1}$  higher than that in December, and

this seasonal concentration difference was further increased to  $225.42 \mu\text{mol kg}^{-1}$  in its downstream. Positive  $\Delta$ [DIC] values occurred mostly in May, indicating that the DIC was higher in the reservoir and downstream of the TGR than upstream (Figure 2b). However, this phenomenon was seasonal, and negative  $\Delta$ [DIC] values occurred mostly in December, indicating that the DIC concentration upstream of the reservoir was higher.

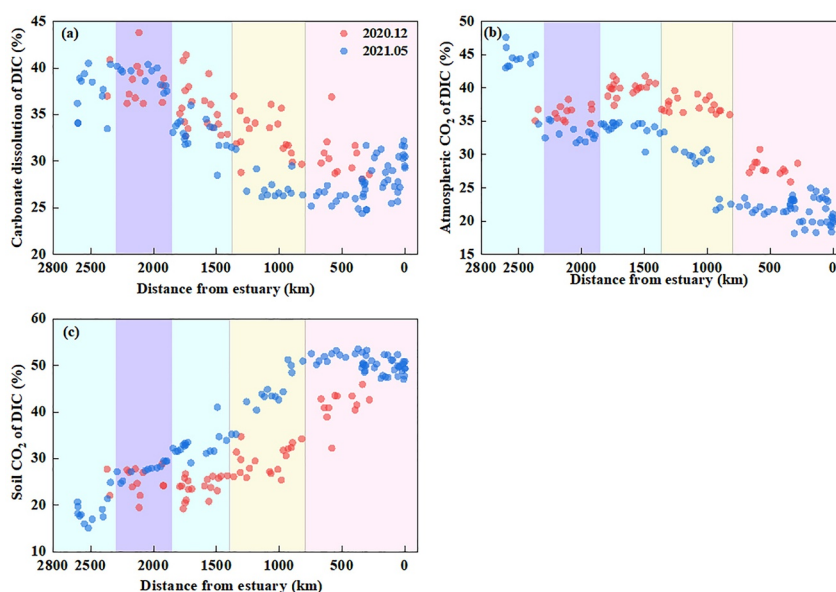
### 3.2. Characteristics of Carbon Isotope and Source of DIC

The  $\delta^{13}\text{C}_{\text{DIC}}$  value exhibited a decreasing trend from upper reach ( $-7.46 \pm 0.52\text{‰}$ ) to lower reach ( $-9.15 \pm 0.41\text{‰}$ ). The  $\delta^{13}\text{C}_{\text{DIC}}$  values between each reach showed significant differences ( $p < 0.05$ ), where values were more enriched in the upper reach and close to the characteristic value of carbonate dissolution ( $-3.9 \pm 0.3\text{‰}$ ), then depleted from the middle to lower reaches and close to the characteristic value of soil  $\text{CO}_2$  ( $-17 \pm 4.0\text{‰}$ ) (Figure 3a; Table S2 in Supporting Information S1). The value of  $\delta^{13}\text{C}_{\text{DIC}}$  exhibited significant seasonality during these two periods (Figure 3a).  $\delta^{13}\text{C}_{\text{DIC}}$  in May was highly depleted ( $-8.57 \pm 0.89\text{‰}$ ) and close to the characteristic value of soil  $\text{CO}_2$  (Figure 3a; Table S2 in Supporting Information S1). In contrast,  $\delta^{13}\text{C}_{\text{DIC}}$  was highly enriched in December ( $-7.85 \pm 0.78\text{‰}$ ) and close to the characteristic value of carbonate dissolution. The values of  $\delta^{13}\text{C}_{\text{DIC}}$  showed a depleted trend from the upstream of TGR to its downstream. The negative value of  $\Delta$  [ $\delta^{13}\text{C}_{\text{DIC}}$ ] occurs May, which means  $\delta^{13}\text{C}_{\text{DIC}}$  was enriched upstream of TGR compared to the reservoir and downstream. Negative  $\Delta$  [ $\delta^{13}\text{C}_{\text{DIC}}$ ] values occurred in May, indicating that the  $\delta^{13}\text{C}_{\text{DIC}}$  was enriched upstream of the TGR compared to in the reservoir and downstream. Notably, although negative  $\Delta$  [ $\delta^{13}\text{C}_{\text{DIC}}$ ] values occurred mostly in December, the  $\delta^{13}\text{C}_{\text{DIC}}$  difference in these areas was smaller than that in May, and the  $\delta^{13}\text{C}_{\text{DIC}}$  at some sample points in the reservoir and downstream of the TGR was even more enriched than that upstream (Figure 3b).



**Figure 3.** Spatiotemporal variations of  $\delta^{13}\text{C}_{\text{DIC}}$  exhibiting depleted trend from upper to lower reach of the Yangtze River. Note that the result of fitting line between distance (from estuary to upper reach) and  $\delta^{13}\text{C}_{\text{DIC}}$  during two periods was  $y = 0.97 \times 10^{-3} \times -9.91$ ,  $R^2 = 0.38$ ; the fitting line in December 2020 was  $y = 1.02 \times 10^{-3} \times -9.50$ ,  $R^2 = 0.87$  and the fitting line in May 2021 was  $y = 0.96 \times 10^{-3} \times -9.59$ ,  $R^2 = 0.85$ . (b) The difference of  $\Delta$  [ $\delta^{13}\text{C}_{\text{DIC}}$ ] between samples in TGR and downstream of TGR. The blue, yellow, and red backgrounds indicated the range of upper, middle, and lower reach, respectively. The extent of TGR was highlighted by purple background.

The source estimation of  $\delta^{13}\text{C}_{\text{DIC}}$  via the Bayesian mixture model exhibited well-defined spatial and seasonal characteristics in the main stream of the Yangtze River. The source proportion of carbonate dissolution and atmospheric  $\text{CO}_2$  both showed a decreasing trend from upper reach to lower reach. In contrary, the source proportion of soil  $\text{CO}_2$  increased from upper reach ( $26.01\% \pm 5.15\%$ ) to lower reach ( $48.63\% \pm 4.25\%$ ). The average contribution of carbonate dissolution to DIC was higher in December ( $34.48\% \pm 3.65\%$ ) than that in May ( $31.09\% \pm 4.82\%$ ), whereas the proportion of soil  $\text{CO}_2$  components decreased from May ( $40.38\% \pm 11.77\%$ ) to December ( $29.34\% \pm 6.99\%$ ) (Figure 4). Atmospheric  $\text{CO}_2$  proportion of DIC decreased from upper to middle reach in December, located near the basin of the Dongting Lake (Figure 4a). In May, the proportion of carbonate dissolution and atmospheric  $\text{CO}_2$  to DIC smoothly decreased from the upper to lower reach, whereas the soil  $\text{CO}_2$  proportion to DIC showed the opposite

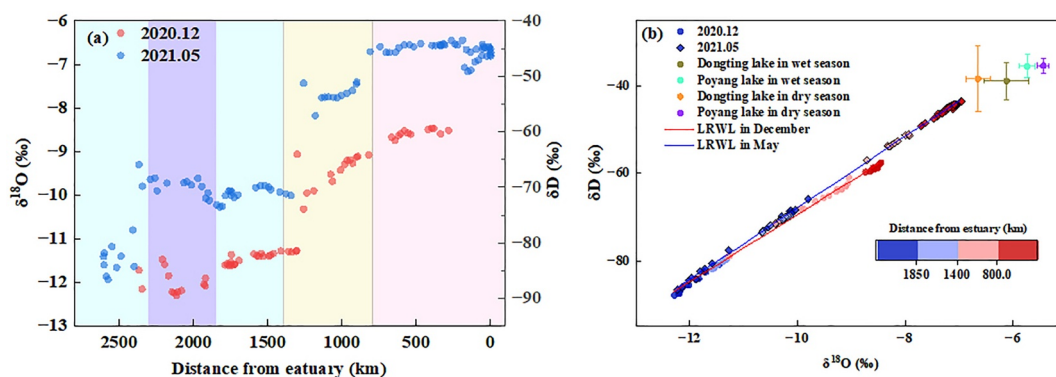


**Figure 4.** Spatiotemporal variation of the contribution proportion of the DIC source. (a) Carbonate dissolution, (b) atmosphere  $\text{CO}_2$  and (c) soil  $\text{CO}_2$ . The blue, yellow, and red backgrounds indicated the range of upper, middle, and lower reach, respectively. The extent of TGR was highlighted by purple background.

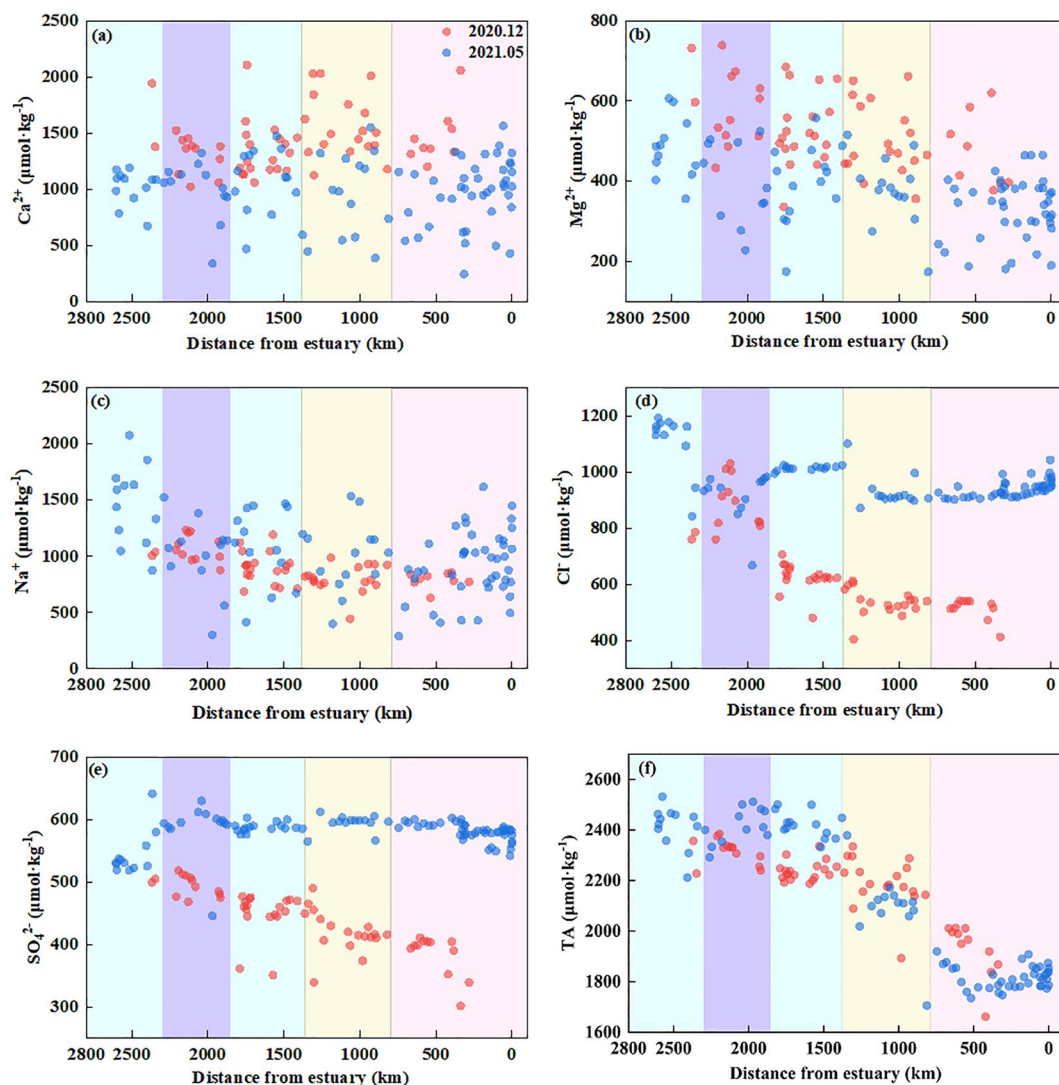
trend. The contribution of soil  $\text{CO}_2$  to DIC increased from the upstream of TGR to its downstream. In addition, the proportion of carbonate dissolution to DIC in December decreased along the upper reaches of the Yangtze River. However, the source of carbonate dissolution in TGR was significantly higher than that in the downstream of TGD in May. The soil  $\text{CO}_2$  proportion of DIC in TGR was higher than that in its upstream and downstream in December.

### 3.3. Hydrological Signals From Upper to Lower Reach

$\delta^{18}\text{O}$  and  $\delta\text{D}$  in the main stream water exhibited synchronous enrichment from the upper to lower reach during the study period. The values of  $\delta^{18}\text{O}$  and  $\delta\text{D}$  between each reach showed significant differences ( $p < 0.05$ ) (Table S2 in Supporting Information S1), such that  $\delta^{18}\text{O}$  in the upper reach ( $-11.13 \pm 0.75\text{‰}$ ) was more depleted than that in the middle reach ( $-9.22 \pm 1.10\text{‰}$ ) and lower reach ( $-7.53 \pm 0.59\text{‰}$ ) (Figure 5a). Notably,  $\delta^{18}\text{O}$  and



**Figure 5.** (a) Spatiotemporal variations of  $\delta^{18}\text{O}$  and  $\delta\text{D}$  exhibiting enriched trend from upper to lower reach of the Yangtze River. The blue, yellow, and red backgrounds indicated the range of upper, middle, and lower reach, respectively. The extent of TGR was highlighted by purple background. (b) Local runoff water line (LRWL) of the Yangtze River.  $\delta^{18}\text{O}$  and  $\delta\text{D}$  in lakes showed more enriched than that in lakes. Red fitting line between  $\delta\text{D}$  and  $\delta^{18}\text{O}$  is data in December 2020 ( $\delta\text{D} = 7.70\delta^{18}\text{O} + 7.59$ ,  $R^2 = 0.99$ ); the blue fitting line is data in May 2021 ( $\delta\text{D} = 8.17\delta^{18}\text{O} + 13.83$ ,  $R^2 = 0.99$ ). The detailed historical data of  $\delta^{18}\text{O}$  and  $\delta\text{D}$  in two lakes in Table S3 of the Supporting Information S1.



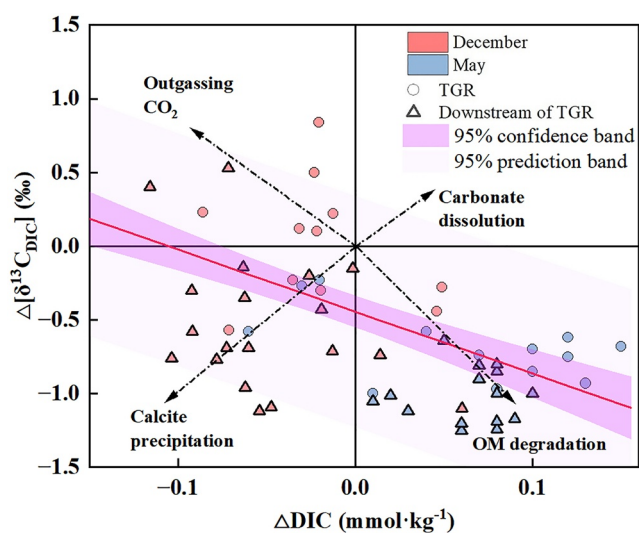
**Figure 6.** Spatiotemporal distribution of major cations and anions in the main stream of the Yangtze River. (a)  $\text{Ca}^{2+}$ ; (b)  $\text{Mg}^{2+}$ ; (c)  $\text{Na}^{+}$ ; (d)  $\text{Cl}^{-}$ ; (e)  $\text{SO}_4^{2-}$ ; (f) TA. The blue, yellow, and red backgrounds indicated the range of upper, middle, and lower reach, respectively. The extent of TGR was highlighted by purple background.

$\delta\text{D}$  in Dongting Lake and Poyang Lake were more enriched than that in the main stream of the Yangtze River during two seasons (Figure 5b). Local runoff water line (LRWL, linear regression of  $\delta\text{D}$  and  $\delta^{18}\text{O}$ ) in the main stream of the Yangtze River showed pronounced seasonality. A higher spatial variation in LRWL was observed in May ( $\text{CV} = 0.19$ ), followed by in December ( $\text{CV} = 0.13$ ). The average rate of change in  $\delta^{18}\text{O}$  and  $\delta\text{D}$  in May (slope = 8.17) was higher than that in December (slope = 7.70). The variation range of  $\delta^{18}\text{O}$  in the middle reach in May ( $-10.42$ – $-7.27\text{‰}$ ) was greater than that in December ( $-11.30$  to  $-9.06\text{‰}$ ) (Figure 5b). The  $\delta^{18}\text{O}$  increased significantly from the upstream of TGR ( $-11.55 \pm 0.75\text{‰}$ ) to the reservoir area ( $-10.25 \pm 0.16\text{‰}$ ). In addition, the LRWL during the two periods was close to the line of the global meteoric water line (GMWL) and the Meteoric Water Line of China, which indicated that water in the main stream of the Yangtze River was mainly from precipitation recharge (Figure S6 in Supporting Information S1).

### 3.4. Major Ion Compositions and TA in the Main Stream

The major cation and anion concentration values followed the order  $\text{Ca}^{2+} > \text{Na}^{+} > \text{Cl}^{-} > \text{Mg}^{2+} > \text{SO}_4^{2-}$ . The concentration of  $\text{Ca}^{2+}$  in December ( $1,439.79 \pm 261.72 \mu\text{mol kg}^{-1}$ ) was higher than that in May ( $1,006.49 \pm 292.49 \mu\text{mol kg}^{-1}$ ) (Figure 6a). The  $\text{Mg}^{2+}$  concentration range from 753.89 to 1,714.63  $\mu\text{mol kg}^{-1}$





**Figure 7.** The relationship between  $\Delta$  [DIC] and  $\Delta$  [ $\delta^{13}\text{C}_{\text{DIC}}$ ] for samples in TGR and its downstream. Note that circles represent samples in TGR and triangles represent samples in downstream of the reservoir area. Samples in different quadrants affected by corresponding process, such as outgassing  $\text{CO}_2$ , calcite precipitation, carbonate dissolution and organic matter degradation.

with mean value of  $434.60 \pm 124.23 \mu\text{mol kg}^{-1}$  in the main stream, which in December was 1.49 times higher than that in May (Figure 6b).  $\text{Na}^+$  in December showed lower variability and concentration ( $893.29 \pm 156.76 \mu\text{mol kg}^{-1}$ ) than that in May ( $1,053.39 \pm 237.18 \mu\text{mol kg}^{-1}$ ) (Figure 6c).  $\text{Cl}^-$  decreased from the upper reach ( $894.00 \pm 188.57 \mu\text{mol kg}^{-1}$ ) to the lower reach ( $848.95 \pm 181.19 \mu\text{mol kg}^{-1}$ ) in the Yangtze River during the two periods, which reflects the decreasing contribution of evaporites from the upper reach to the lower reach (Figure 6d).  $\text{SO}_4^{2-}$  in December decreased from the middle to lower reach, and in May, it varied smoothly in the entire basin with a mean of  $581.935 \pm 23.44 \mu\text{mol kg}^{-1}$  (Figure 6e).

The TA values were significantly different ( $p < 0.05$ ) in the three reaches during the sampling periods. TA values were high in the upper reaches ( $2,351.31 \pm 96.53 \mu\text{mol kg}^{-1}$ ), then decreased from the middle reaches ( $2,158.22 \pm 134.55 \mu\text{mol kg}^{-1}$ ) to the lower reaches ( $1,842.60 \pm 77.66 \mu\text{mol kg}^{-1}$ ) (Figure 6f). The mean value of TA in December was  $2,179.36 \pm 154.26 \mu\text{mol kg}^{-1}$  and reached a lower value in May ( $2,115.14 \pm 287.66 \mu\text{mol kg}^{-1}$ ) (Figure 6f). The temporal and spatial variation trend of alkalinity was consistent with DIC, which indicated that the biogeochemical processes of DIC and TA were regulated by TGR and two large lakes (Figures 2 and 6f).

## 4. Discussion

### 4.1. Dam Operation Constrains Riverine DIC Concentration and Source

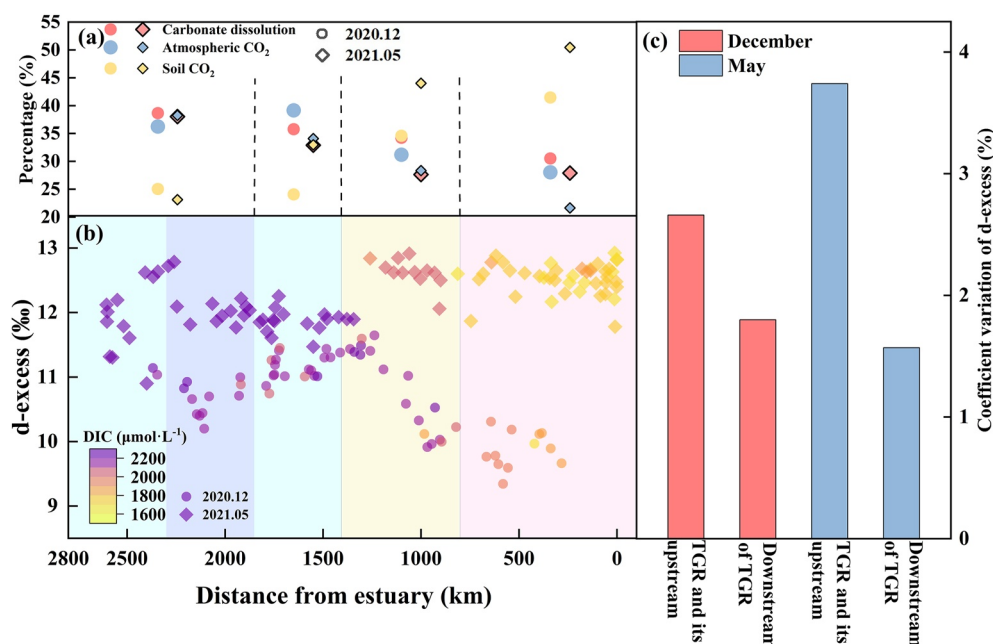
#### 4.1.1. Outgassing $\text{CO}_2$ and Calcite Precipitation Process in Dry Season

The outgassing of  $\text{CO}_2$  and calcite precipitation were the main factors affecting the lower DIC concentrations in the reservoir and its downstream area in December (Figure 7). The flux of sediments reached  $1.07 \times 10^6 \text{ t yr}^{-1}$  in the TGR (Li et al., 2021). Interception of terrestrial and autochthonous organic matter by dams regulated the DIC cycle. With the reduction in phytoplankton photosynthesis in December, mineralization of resuspended sedimentary organic matter further yielded  $\text{CO}_2$  to reduce the pH and DO of surface water in TGR (Liu et al., 2016) (Figures S7c and S7e in Supporting Information S1). This process increased the soil  $\text{CO}_2$  proportion of DIC in the reservoir area. The  $p\text{CO}_2$  in the reservoir decreased from depth of 18 m to surface water in December, further indicating the process of escaping the accumulated  $\text{CO}_2$  from reservoir bottom to the surface (Figure S8a in Supporting Information S1). Increased  $\text{CO}_2$  emissions from degassing cause DIC consumption in the reservoir area (Li et al., 2018). Notably,  $p\text{CO}_2$  downstream of TGR increased from the depth to the surface, and its value was lower than that in the reservoir (Figure S9 in Supporting Information S1). The similar water column temperatures indicated that temperature was not the main reason for the difference of  $p\text{CO}_2$  (17.50 and 17.66°C). This phenomenon may have been caused by the reduction in sediment because POC and DOC concentrations decreased from the deep water to surface water downstream of the dam (Figures S8b and S8c in Supporting Information S1). The reduction of organic matter weakens the mineralization downstream of TGR. Higher pH and lower  $p\text{CO}_2$  facilitated calcite precipitation (Equation 3). The significant correlation between  $\Delta[\text{Ca}^{2+}]$  and  $\Delta[\text{DIC}]$  downstream of TGR ( $r^2 = 0.44$ ) in December reveals that the DIC decreased due to calcite precipitation (Figure S10 in Supporting Information S1).



#### 4.1.2. Organic Matter Degradation and Retention Effect in Wet Season

Abundant precipitation in the wet season led to a continuous flow of terrestrial organic matter into the TGR. The degradation of organic matter increased the soil  $\text{CO}_2$  proportion of DIC and contributed to depleting the  $\delta^{13}\text{C}_{\text{DIC}}$  (Figure 7). Although the degradation of organic matter may increase dissolved  $\text{CO}_2$  in water, the negative correlation between Chl *a* and dissolved  $\text{CO}_2$  ( $r = -0.49$ ,  $p < 0.01$ ) in the reservoir area in May highlighted how  $\text{CO}_2$  outgassing was weakened by phytoplankton photosynthesis (Figure S11 in Supporting Information S1) (Pu



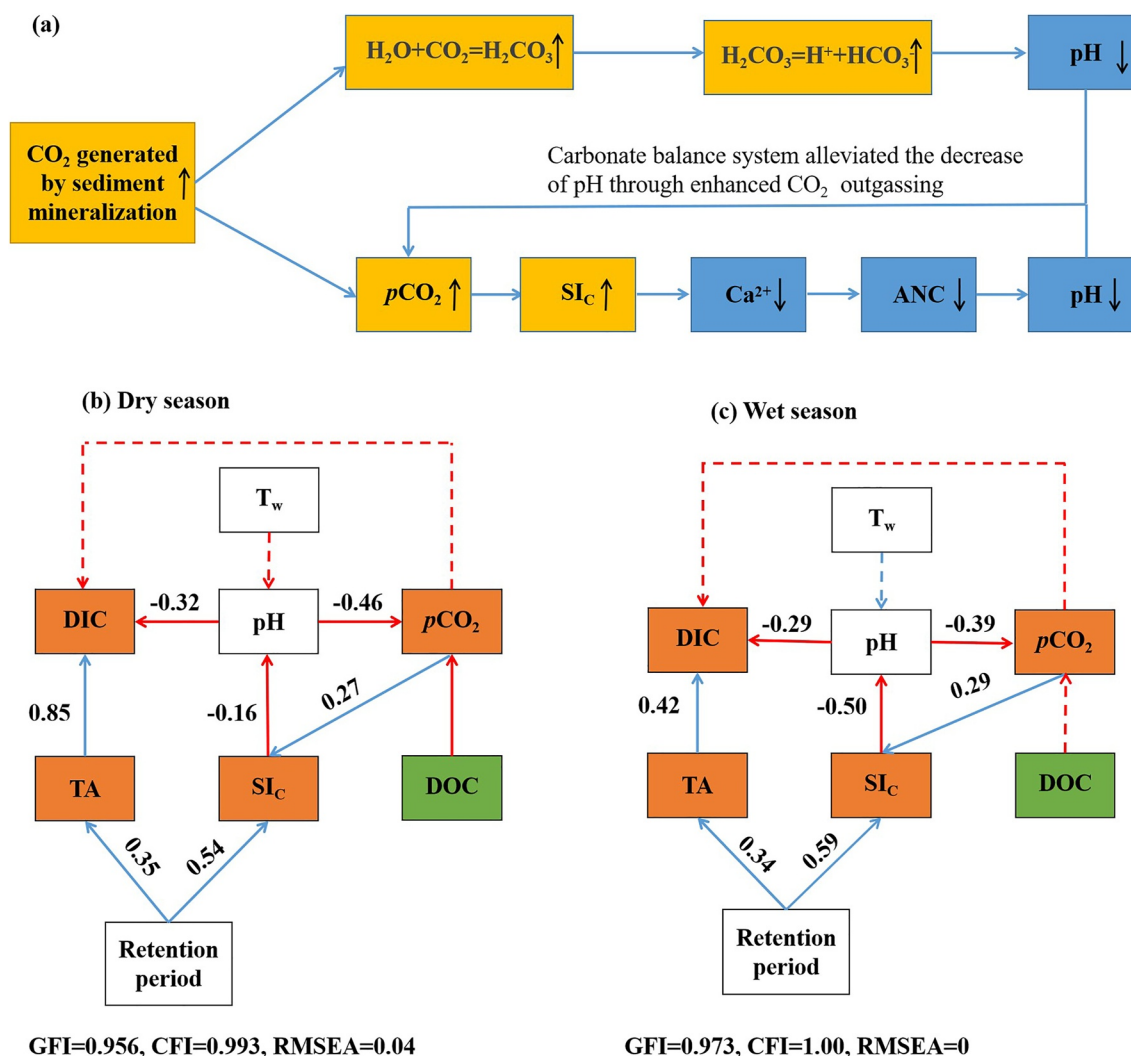
**Figure 8.** (a) DIC source proportion in different reaches; (b) The value of d-excess in the main stream of the Yangtze River; (c) Coefficient variation d-excess of TGD and its upstream versus downstream. According to the geographical location, purple boxes are used to identify the extent of the reservoir area and distinguish its upstream and downstream. Blue, yellow and red shaded areas indicate upper, middle, and lower reaches in this study, respectively.

et al., 2017; Uthicke & Fabricius, 2012). In addition, dam impoundment increased the DIC and its carbonate dissolution proportion in May (Figure 8a). Longer residence time of water in reservoirs induced more vigorous mixing, contributing to relatively smaller fluctuations in the isotopic signatures and manifesting as lower CV of d-excess (Figure 8b) (Deng et al., 2016; Wu et al., 2022). Based on the monthly storage change and outbound flow, the assessment results indicated that the drainage period of the reservoir in May was 3.38 times longer than that in December (Text S5 in Supporting Information S1). Longer retention time contributed to the lower CV of d-excess in May (1.5%) than that in December (1.7%) (Figure 8c). The extended period of water retention in the reservoir in May decreased the runoff flow. The sufficient time for water–rock reactions to occur due to the extended retention time increased the carbonate dissolution proportion and DIC concentration in the TGD and its upstream area in May (Figures 2a and 3a) (Han et al., 2018; Wu & Firoozabadi, 2010).

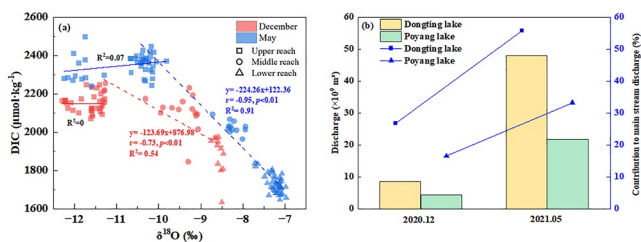
SEM revealed the feedback mechanism of the carbonate balance system to continuous saturation of CO<sub>2</sub> driven by sediment mineralization in the reservoir water from 2009 to 2021 (TGD in the normal stage). Mineralization of the accumulated sediment yields saturated *p*CO<sub>2</sub> in the reservoir water, thereby promoting calcite precipitation and reducing the acid neutralization capacity of water (Figure S12 in Supporting Information S1). The carbonate buffer system alleviated the decrease of pH by enhancing the outgassing of CO<sub>2</sub>, which further formed a positive feedback loop of water acidification in the reservoir area (Figure 9a). Previous studies indicated that more acidified conditions would promote carbonate dissolution and result in an increase in TA and HCO<sub>3</sub><sup>-</sup> (Kapetanaki et al., 2018, 2020). Notably, the SEM results show that *p*CO<sub>2</sub> variation has no significant impact on DIC concentration (*p* > 0.05), which can be attributed to the interannual variation of DIC in TGD driven by HCO<sub>3</sub><sup>-</sup> generated by mineralization and water acidification, rather than the loss of DIC caused by CO<sub>2</sub> outgassing (Figures 9b and 9c). Thus, sediment mineralization increased annual *p*CO<sub>2</sub> and DIC concentrations in the TGR.

#### 4.2. River-Communicating Lake Recharge Regulates DIC Concentrations and Sources

The significant negative correlation between DIC concentration and δ<sup>18</sup>O only occurred downstream of Dongting and Poyang lake; thus, the flow of lakes mixed into main stream enriches hydrological signals and contributes to the dilution effect of DIC (Figure 10a). Wide surface area and slow runoff make water in lakes suffer intense sub-cloud secondary evaporation, thus more enriched deuterium and oxygen signal formed in lakes than that in main stream (Deng et al., 2016; Ding et al., 2014; Li, Song, et al., 2022). Isotopic source analysis (δD and



**Figure 9.** The impact of environmental factors on DIC in the reservoir area (a) feedback mechanism of carbonate balance system response to continuous saturation of CO<sub>2</sub>, yellow boxes represent increasing trends and blue represent decreasing trends; (b) dry season; (c) wet season. In SEM model, blue and red solid lines indicate the significant positive and negative correlation paths ( $p < 0.05$ ), solid and dashed lines suggest significant ( $p < 0.05$ ) and not significant correlation ( $p > 0.05$ ). Orange boxes represent inorganic system and the green represent organic carbon. Some of data were compiled from (Wu and Han (2018), Heath et al. (2020), Li et al. (2020a, 2020b), detailed in Table S4 of the Supporting Information S1), and other data were collected from this study.



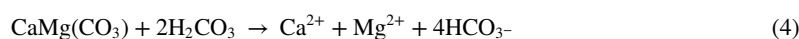
**Figure 10.** (a) Relationships between DIC and δ<sup>18</sup>O during the two periods in the main stream of the Yangtze River, the squares, circles, and triangles symbols indicated the range of upper, middle, and lower reach, respectively; (b) Contribution proportion of lakes recharge to discharge of main stream, the contribution percentage is calculated using Bayesian mixing model and end member of δ<sup>18</sup>O are samples near lakes-river confluence and historical value of two lakes in Figure 5b.

δ<sup>18</sup>O) showed that the contribution of Dongting Lake and Poyang Lake to the main stream discharge in December was 26.7% and 16.5%, compared with 55.7% and 33.1% in May. These results are consistent with the changes in lake discharge monitored by the hydrological stations (Figure 10b). Therefore, lake recharge was the main cause of the variation in the hydrological signals and DIC in the middle and lower reaches. The results of ion stoichiometry and carbon isotope analyses further explained the spatial variation in the DIC. Carbonate weathering by CO<sub>2</sub>(aq) gradually weakened in the lake inflow area, which led to a decreasing trend in the DIC, carbonate dissolution, and atmospheric CO<sub>2</sub> proportions from the middle to lower reaches (Figure 8). Previous studies have revealed that carbonate weathering weakens from the middle to the lower reaches because of the lower abundance of carbonate rocks, which is consistent with the results of this study (Figure S1 in Supporting Information S1) (Zachara et al., 2016; Zhang et al., 2014). In

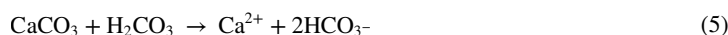
addition, river-communicating lakes are considered terrestrial carbon sinks that export terrigenous substances to increase the soil CO<sub>2</sub> proportion of DIC (Cao et al., 2016; Lu et al., 2018).

Seasonal differences in the DIC concentration and source proportion in the middle and lower reaches are regulated by lake recharge in the Yangtze River (Chetelat et al., 2008; Shan et al., 2021). The higher contribution of lake recharge to the main stream and the better fitting relationship between the DIC and δ<sup>18</sup>O ( $R^2 = 0.91$ ) indicated that the DIC was more influenced by lake contributions to discharge in May than in December (Figures 10a and 10b). Lakes recharge diluted the TA and reduced the contribution of carbonate dissolution to DIC, which result in lower DIC in the middle and lower reaches in May (Figures 4a and 6f). Notably, the concentrations of Ca<sup>2+</sup> and Mg<sup>2+</sup> in the middle and lower reaches in December were higher than those of HCO<sub>3</sub><sup>-</sup>, which may imply that (a) carbonate dissolution involves acids other than carbonic acid, and (b) evaporite dissolution provides extra sources of Ca and/or Mg. The addition of SO<sub>4</sub><sup>2-</sup> can lead to a balance in the anions and cations, which indicates that additional sulfuric acid is dissolved with the carbonate to increase the DIC concentration in December (Figure 11b) (Jia et al., 2021). Precipitation research in the Yangtze River Basin showed little difference in SO<sub>4</sub><sup>2-</sup> in rainfall in different seasons, which was much lower than in the river (Li et al., 2020a, 2020b; Yoon et al., 2008). Therefore, acid deposition caused by rainfall may not be a significant contributor to sulfuric acid in the middle and lower reaches. In contrast, the better correlation between metal cations (Na<sup>+</sup> and Mg<sup>2+</sup>) and SO<sub>4</sub><sup>2-</sup> in the water samples suggests that evaporation and dissolution of minerals such as mirabilite (Na<sub>2</sub>SO<sub>4</sub>·10H<sub>2</sub>O) and polyhalite (MgSO<sub>4</sub>) were an additional sulfate source in December (Figures 11c and 11d) (Wang et al., 2022; Zhang et al., 2021). Abundant evaporite reserves in the two lake basins and the characteristics of evaporation, that is, easy crystallization in dry season, further support this hypothesis, as additional sulfuric acid increased the carbonate dissolution proportion of DIC in December (Chetelat et al., 2008). In addition, congruent calcite dissolution in the middle reach in December increased carbonate dissolution and produced more HCO<sub>3</sub><sup>-</sup> than that in May (Figure 11e). Previous studies have shown that calcite has a faster dissolution rate than dolomite (Fairchild et al., 2000; Nash et al., 2013; Yadav et al., 2008), which further increased the seasonal difference in DIC in the middle reach (Equations 4 and 5).

Congruent dolomite dissolution:



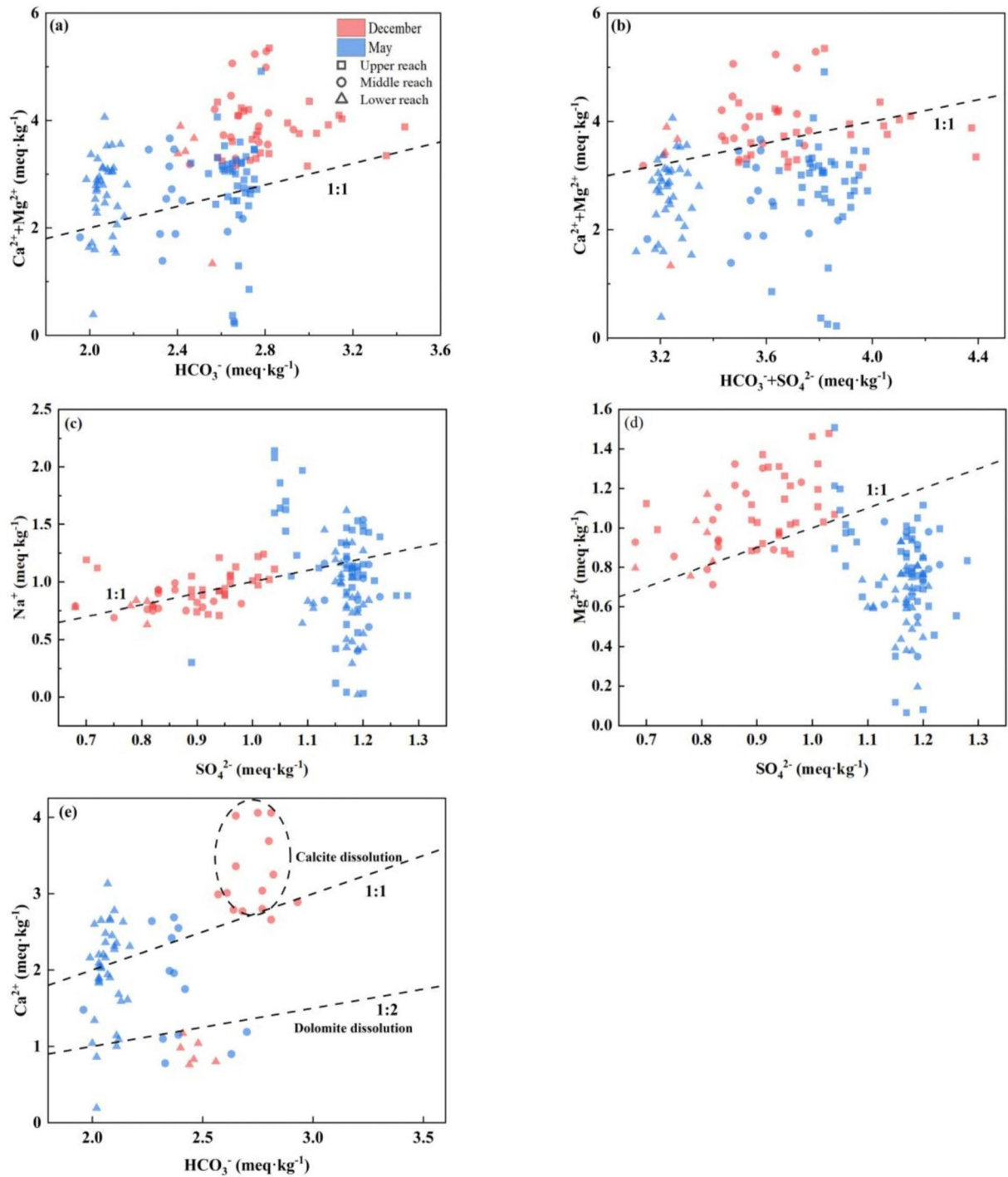
Congruent calcite dissolution:



### 4.3. Effects of Environmental Factors on DIC and Its Sources

A significant correlation between environmental factors and DIC has been widely reported in global rivers (Kaijser et al., 2021; Liu et al., 2021). Redundancy analysis (RDA) showed that DIC in the upper reach of the Yangtze River mainly originated from carbonate dissolution and rock weathering with atmospheric CO<sub>2</sub> (Figure 12) (Chetelat et al., 2008; Zhang et al., 2014). Moreover, a positive correlation between the soil CO<sub>2</sub> proportion of DIC and organic carbon (OC, POC, and DOC) appeared in the RDA1 negative axis, where water samples were distributed in the lower reach (Figure 12; Table S7 in Supporting Information S1). This phenomenon indicates that the mineralization of organic matter increases the soil CO<sub>2</sub> proportion of DIC in the lower reach, and this effect is also evidenced by the significant positive correlation between OC and *p*CO<sub>2</sub> ( $p < 0.01$ ) (Figure S13 in Supporting Information S1) (Dutta et al., 2019; Zhong et al., 2018). As the DOC concentration in the main stream of the Yangtze River has been increasing over the years, the variation in DIC concentration and source caused by OC transformation needs more attention. Moreover, higher water temperatures often enhance mineral weathering and promote DIC production (Liu et al., 2020; Prokushkin et al., 2011), but there was a significant negative correlation between water temperature and DIC in this study ( $p < 0.01$ ) (Figure S13 in Supporting Information S1). This indicates that the variation in DIC is more sensitive to dam operation and lake recharge than to the warming effect on rock weathering. DIC accounts for 89.47% of total carbon, 93.48% of TDC, and 98.89% of inorganic carbon on average in this study. Mean DIC concentrations were higher after dam operation than before, which further highlights the disturbance of dam and river-lake exchange on watershed carbon cycle (Figure S14 in Supporting Information S1).





**Figure 11.** Stoichiometric relation among major ions in the main stream of the Yangtze River. (a)  $\text{Ca}^{2+} + \text{Mg}^{2+}$  versus  $\text{HCO}_3^-$ ; (b)  $\text{Ca}^{2+} + \text{Mg}^{2+}$  versus  $\text{HCO}_3^- + \text{SO}_4^{2-}$ ; (c)  $\text{Na}^+$  versus  $\text{SO}_4^{2-}$ ; (d)  $\text{Mg}^{2+}$  versus  $\text{SO}_4^{2-}$ ; (e)  $\text{Ca}^{2+}$  versus  $\text{HCO}_3^-$ .

#### 4.4. Impact of Dam on C-Q Relationship in Yangtze River and Other Rivers

High discharge often shortens the runoff transit time and depletes the DIC source supply (Liu et al., 2020; Wang et al., 2013; Zhong et al., 2021b). As with most Chinese rivers, previous studies have revealed that DIC concentrations ( $C$ ) decreased with increasing discharge ( $Q$ ) in the Yangtze River ( $C$ - $Q$  relationships:  $C = aQ^b$ , where  $a$  and  $b$  are fitted parameters,  $b$  values near 0 indicate the chemostatic behavior of DIC with changing discharge)



climate and basalt bedrock (Eiriksdottir et al., 2017). When dams release water and increase the discharge scour of the basalt bedrock, the shortened fluid transit time contributes to the depletion of DIC; thus, the DIC concentration is diluted by an increase in discharge after dam operation (the  $b$  value becomes negative) (Figure 13d). Da River, Lo River, and Thao River are located in the Red River basin of Vietnam, which has a tropical monsoon climate and carbonatite bedrock (Phuong et al., 2018). Carbonate weathering is the main source of  $\text{HCO}_3^-$ , and the higher discharge intensifies chemical weathering by increasing the reactive surface area. Therefore, the DIC showed the enrichment behavior with increasing discharge after dam operation, which differed from that of other rivers ( $b > 0$ ) (Figure 13d). These results highlight that the geology of the watershed and damming change the chemostatic behavior of  $C-Q$ . With the increasing number of dams built worldwide and the frequent occurrence of extreme weather caused by global warming, the transport conditions of solutes will change greatly (Castello & Macedo, 2016; Haddeland et al., 2014), in watersheds with widespread minerals with fast carbonate weathering kinetics, the DIC will increase with increasing discharge. This study provides a framework for better understanding the transmission and transformation processes of DIC in river–reservoir systems.

## 5. Conclusion

In this study, isotope tracing and hydrochemical analysis methods were used to interpret the influence of dam operation and lake recharge on DIC. DIC in the reservoir area reached  $2,269.94 \mu\text{mol kg}^{-1}$  and then decreased by 7.77% and 21.95% after the inflow of Dongting Lake and Poyang Lake, respectively, into the main stream.  $\delta^{13}\text{C}_{\text{DIC}}$  analyzed using Bayesian mixture model indicated that DIC originates from rock weathering (involved carbonate dissolution and atmospheric  $\text{CO}_2$ ) and soil  $\text{CO}_2$  generated by organic matter mineralization. Lake recharge weakened the rock weathering process such that the carbonate dissolution and atmospheric  $\text{CO}_2$  proportions of DIC decreased from the upper reach to the lower reach. Meanwhile, terrestrial substance transported from lakes increased the soil  $\text{CO}_2$  proportion of DIC in middle and lower reach of the Yangtze River. The regulation of TGD on DIC was found to be seasonally independent in upper reach of the Yangtze River. During the dry season, the DIC decreases in the TGD area, dominated by  $\text{CO}_2$  degassing and calcite precipitation. In the wet season, a lower CV of d-excess revealed a longer runoff retention time in the reservoir, leading to sufficient water–rock interactions to increase the DIC and the proportion of DIC from carbonate dissolution. Enrichment of  $\delta^{18}\text{O}$  from the middle to lower reaches revealed the flow of lakes mixed into the main stream of the Yangtze River, which decreased the DIC concentration and the source proportion of carbonate dissolution. Globally, the impact of damming on the DIC transmission process was evaluated among the Yangtze River and other rivers, and damming in watersheds with different geologies has limited the transport and source of DIC, thereby further altering the original chemostatic behavior of  $C-Q$ . Future studies should focus on the effects of damming and lake recharge on chemical weathering in watersheds. Research on chemical weathering processes under a variety of geologies is also needed to better understand the transmission and transformation processes of DIC in the river–ocean continuum.

## Conflict of Interest

The authors declare no conflicts of interest relevant to this study.

## Data Availability Statement

**Data Citation:** The original data related to this article about water chemical parameters in the Yangtze River are available on the website of Zenodo (Zhao, 2023, <https://doi.org/10.5281/zenodo.8273583>). **Software Citation:** Stock (2020), <https://rdocumentation.org/packages/MixSIAR/versions/3.1.12>.

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